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by

R. A. Sacks, H. K. Geyer, S. J. Grammel, and E. D. Doss

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MODIFIED NASA-LEWIS CHEMICAL EQUILIBRIUM CODE FOR MHD APPLICATIONS

by

R. A. Sacks, H. K. Geyer, S. J. Grammel, and E. D. Doss

Engineering Division

December 1979

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ABSTRACT

A substantially modified version of the NASA-Lewis Chemical Equilibrium Code (1) has recently been developed. The modifications were designed to extend the power and convenience of the Code as a tool for performing comtustor analysis for MHD systems studies. This report describes the effect of the programming details from a user point of view, but does not describe the Code in detail.

I. INTRODUCTION

The NASA-Lewis Chemical Equilbrium Code [1] had gained wide acceptance as a valuable and accurate tool for computing thermodynamic properties and chemical equilibrium compositions. In a companion publication [2], two of the authors have described a preprocessor program (GPSAP) written to facilitate general systems analysis studies. By adapting the NASA code to the format that GPSAP requires for component models, the very powerful systems analysis capabilities are obtained (e.g., execution time parameter alterations, automated sensitivity studies, and optimization features).

Several additional capabilities were also added to the new code, to implement or simplify combustor studies, especially as related to MHD applications. These new features include:

- 1) a routine for computing the electrical properties of the equilibrium mixture.
- 2) a shorthand data input option for treating standard hydrocarbon fuels.
- 3) a facility for incorporating heat losses (as a fraction of thermal input) into the NASA HP mode computations, and
- 4) a standard printed summary of equilibrium thermodynamic properties.

Section II describes the modifications that have been made. Section III is a users guide, detailing the input data and giving information necessary for efficient systems studies. Two examples, both demonstrating the power of the modified code and illustrating its use are presented in Section IV. Throughout, it is assumed that the reader is familiar with both the original version of the NASA code and the fundamentals of GPSAP usage.

II. CODE MODIFICATIONS

A. GPSAP Compatibility

As described in Ref. 2, GPSAP requires certain standard features from the component models. Specifically, it is necessary that the data input, calculational, and data output modules be clearly separated entities, each with its own entry point, and that there are appropriate facilities for communicating data between the main program and the calculational block.

Although it is quite generally possible to write PL/1 drivers properly modularized for FORTRAN models, the coupling is most easily accomplished with PL/1 models. The NASA code, as originally written, did not permit alteration of the input data at execution time. The most straightforward approach to remedying this situation was to translate into PL/1 the routines for data input and reduction, output, and the calculational driver, and to rearrange the program logic according to GPSAP requirements. In doing so, the rocket, detonation, and constant volume equilibrium modes were discarded for the sake of execution efficiency. Aside from eliminating these options, the equilibrium calculations (EQLBRM, MATRIX, GAUSS, and CPHS) were left unaltered in FORTRAN.

The three PL/1 entry points to the new version are,

NASAIN, NASA, NASAOUT.

NASAIN simply reads the input data from file CARDIN and stores it.

NASA performs any desired alterations to the input, reduces the resulting set of data to array form, calls EQLBRM repeatedly to perform the requested tasks, and, if instructed, prints various amounts of intermediate output.

NASAOUT prints the summary of thermodynamic and electrical properties and, if instructed, stores the same on disk file FITDAT for post processing.

As in all GPSAP modes, the different entry points are now completely independent. Thus, from a single call to NASAIN, NASA can be (and typically is) called repeatedly, the input data being altered each time from MAIN. The power of the approach derives from the fact that those input data alterations to one NASA call can (and typically do) derive from the results of a previous call. Thus, parameter optimizations may be performed in a single computer job. It has been extensively verified that, for a given set of input data, the answers obtained with the new code are identical to those obtained previously.

B. Electrical Conductivity Procedure

1. Model Description

The plasma electrical conductivity and mobility are computed by use of the algorithm of Demetriades and Argyropoulos [3]. This method calculates the electrical conductivity, using Grad's 13-moment approximation to derive a generalized Ohm's law for a multicomponent plasma. The electrical properties of the plasma are obtained in two successive approximations that are equivalent to the first and second approximations in the expansion of Sonine polynomials in the Chapman-Enskog approach.

In this method, the coefficients of Ohm's law (electrical properties of plasma) are computed if three electron-neutral parameters are known as functions of electron temperature [4]: The effective collision cross section for momentum transfer, Q_{ek} , and the two weighting factors, $A_{ek}^{(2)}$ and $A_{ek}^{(5)}$, that characterize the electron-neutral interaction.

The equations used in the calculation of the electrical properties of plasma that are required in Ohm's law are as follows:

Ohm's law: E + U x B =
$$\left(\frac{1}{\sigma}\right)$$
J + χ J x B

where the coefficients, σ and x, are given by

$$\sigma_0 = e^2 n_e/m_e vt$$

$$\sigma = \sigma_0/(1-\Delta)$$

$$x = (e n_e)^{-1}$$

or $\Lambda_{ee} = 8.7592 \cdot 10^6 T_e^{3/2} / n_e^{1/2}$

where

$$vt = \sum_{k \neq e} \tau_{e,k}^{-1}$$

$$\tau_{e,k}^{-1} = \frac{4}{3} n_{k} (8k T_{e}/m_{e})^{1/2} Q_{ek}$$

$$\Delta = \frac{5}{2} v_{o}^{2} \tau_{e}^{*} / v_{t}$$

$$v_{o} = \sum_{k \neq e} A_{ek}^{(2)} \tau_{e,k}^{-1}$$

$$(\tau_{e}^{*})^{-1} = \frac{2}{5} [1 - (2 \ln \Lambda_{ee})^{-1}] \tau_{ee}^{-1} + \sum_{k \neq e} A_{ek}^{(5)} \tau_{e,k}^{-1}$$

$$\Lambda_{ee} = 3\lambda (4\pi\epsilon_{o}/e^{2}) kT$$

$$\lambda^{-2} = \frac{1}{\epsilon_{o}} \sum_{s} \frac{n_{s} e_{s}^{2}}{k T_{s}}$$

These equations are the ones used in Ref. 3. However, in Ohm's law, the effective electric field term resulting from electron temperature and pressure gradients is neglected. Also, the effect of magnetic field on the Hall field coefficient (x) is neglected; it will be discussed later.

2. Input Data

In the computation of the electrical properties of the plasma, the following species are considered: OH, Cs, Ar, CO, CO₂, N₂, O₂, H₂O, NO, KOH, SO₂, K, and e (electrons). However, the NASA code considered all the possible species for the particular fuel used. The momentum cross sections, Q's, and the weighting factors, $A^{(2)}$'s and $A^{(5)}$'s, for the specie considered are tabulated as functions of plasma temperature and are given in Table I. The values follow closely the data given by Spencer and Phelps.[5]. However, these data are subject to change should new experimental results become available.

For the calculation of the electrical properties of plasma at other temperatures than those given in the table, linear interpolation is used.

Table 1

O's, A(2)'s and A(5)'s (as functions of temperature from 1700 to 3000K)

```
OH
     17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
     24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2, 87.00, 82.19, 77.90, 74.05, 70.58, 67.42, 64.54, 61.91, 59.48, 57.25, 55.18, 53.26, 51.48, 49.81,
     -. 1981, -. 1969, -. 1958, -. 1948, -. 1939, -. 1930, -. 1922,
    -.1914, -.1908, -.1901, -.1895, -.1889, -.1883, -.1878,
     "921,
                      .920,
             .920,
                             .920, .919, .919,
     .918,
             .917.
                      .917,
                               .917,
                                      .916,
                                               . 916,
                                                         .916,
<u>Cs</u>
     17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2, 24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
    685.48, 645.31, 607.81, 572.99, 540.78, 511.07, 483.71,
    458.56, 435.45, 414.24, 394.77, 376.91, 360.50, 345.44,
    -.2116, -.2334, -.2520, -.2677, -.2806, -.2909, -.2989, -.3046, -.308, -.3100, -.3101, -.3085, -.3055, -.3012,
    .501,
              .539,
                        .582,
                                  .626,
                                            .673,
                                                      .720,
                                                                 .768,
    .815,
              .862,
                        .907.
                                  .951,
                                            .994,
                                                      1.034.
                                                                1.073.
<u>Ar</u>
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
    24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
            .526, .560, .596, .632, .669,
    .494,
                                                         .707.
    .745; .783, .821, .860, .899, .937, .976, .6356, .6574, .6723, .6822, .6882, .6913, .6924,
    .6919, .6904, .6880, .6851, .6819, .6784, .6749,
    3.085, 3.062, 3.033, 3.000, 2.965, 2.930, 2.896,
    2.863, 2.832, 2.803, 2.775, 2.750, 2.727, 2.705,
<u>C0</u>
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
    24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
    13.050, 13.360, 13.669, 13.979, 14.290, 14.601, 14.911,
             15.523, 15.823, 16.117, 16.403, 16.680, 16.947,
    15.219,
    .3617,
              .3568,
                        .3722,
                                 . 3777,
                                            .3823,
                                                      .3872,
                                                                .3908,
    .3933,
              .3948,
                        .3951,
                                  .3943,
                                            .3924,
                                                      .3894,
                                                                .3855,
    1.771,
              1.800,
                        1.230,
                                 1.840,
                                            1.850,
                                                      1.851,
                                                                1.844,
    1.829,
              1.807,
                        1.779,
                                 1.746,
                                           1.708,
                                                     1.668,
<u>CO2</u>
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
    24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
                                 16.39, 15.64,
    19.15,
                       17.22,
             18.13,
                                                    14.96,
                                                                14.34,
             13.25,
    13.77,
                        12.77,
                                 12.33,
                                           11.92,
                                                     11.55,
            -.1823, -.1836, -.1841, -.1840, -.1831,
   -. 1802,
                                                               -.1816,
   -.1796, -.1769, -.1738, -.1700, -.1658, -.1610, -.1558, .857, .870, .883, .897, .911, .926, .942,
    .957,
              .974,
                       .992,
                                 1.010,
                                           1.029,
                                                     1.049,
```

```
12
    17. E2, 18. E2, 19. E2, 20. E2, 21. E2, 22. E2, 23. E2,
    24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.F2, 30.E2, 9.080, 9.178, 9.270, 9.358, 9.445, 9.532, 9.620,
    9.710, 9.804, 9.903, 10.007, 10.115, 10.229, 10.349,
    .2757, .2741, .2737, .2747, .2769, .2804, .2852,
    .2911, .2980, .3058, .3143, .3234, .3330, .3427,
    1.428, 1.444, 1.467, 1.495, 1.529, 1.568, 1.611,
    1.657, 1.705, 1.753, 1.801, 1.847, 1.891, 1.931,
<u>02</u>
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
    24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
    5.624, 5.696, 5.765, 5.832, 5.896, 5.958, 6.018,
    6.075, 6.129, 6.181, 6.231, 6.278, 6.322, 6.364,
    .2896, .2896, .2897, .2898, .2898, .2895, .2890,
    .2881, .2869, .2855, .2837, .2816, .2793, .2768, 1.498, 1.501, 1.502, 1.501, 1.496, 1.490, 1.482,
    1.472, 1.460, 1.447, 1.433, 1.419, 1.404, 1.388,
<u>H20</u>
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2, 24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
    108.532, 102.550, 97.229, 92.443, 88.120, 84.197, 80.620,
   77.344, 74.333, 71.555, 68.985, 66.598, 64.377, 62.305,
   -.1968, -.1955, -.1942, -.1931, -.1921, -.1911, -.1903,
             -.1887, -.1881, -.1874, -.1869, -.1863, -.1859,
   -.1895,
    .923,
             .923,
                      .922,
                                .922,
                                         .921,
                                                   .920,
                                                             .919,
    .919,
             .918,
                       .917,
                                .916,
                                         .916,
                                                   .915,
                                                             .915,
NO
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
   24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.F2, 30.E2,
    16.592, 17.05, 17.43, 17.735, 17.97, 18.15, 18.28,
   18.369,18.42, 18.44, 18.43, 18.39, 18.34, 18.27,
   .4080, .3766, .3478, .3212, .2969, .2745, .2539,
   .2351, .2177, .2017, .1869, .1734, .1608, .1493,
   1.269, 1.19, 1.124, 1.071, 1.027, .991,
                                                     .962.
    .938,
            .919.
                  .904, .892, .883,
                                             .876.
                                                     .871.
KOH
    17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2,
   24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2,
   620.30, 585.90, 555.12, 527.42, 502.37, 479.59, 458.79, 439.72, 433.17, 405.98, 390.97, 377.03, 364.05, 351.93,
   ···· 1994, -.1993, -.1922, -.1990, -.1990, -.1990, -.1990,
            -.1991, -.1991, -.1992, -.1993, -.1994, -.1995, .903, .902, .901, .901, .900, .900,
   -.1990,
   .903,
   .899,
             .898,
                      .898,
                                .898.
                                                   .897,
                                         .897,
```

```
<u>502</u>
           17.E2, 18.E2, 19.E2, 20.E2, 21.F2, 22.F2, 23.E2, 24.E2, 25.F2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2, 84.93, 80.25, 76.08, 72.33, 68.95, 65.88,
                                   58.15, 55.98, 53.97, 52.098, 50.359, 48.736, -.1957, -.1945, -.1934, -.1923, -.1914, -.1905, -.1890, -.1884, -.1878, -.1872, -.1867, -.1863, .923, .922, .922, .921, .920, .919,
           60.51,
           -.1897,
           .923,
           .918,
                                       .917.
                                                                                               .916,
                                                                                                                          .915,
K
          17.E2, 18.E2, 19.E2, 20.E2, 21.E2, 22.E2, 23.E2, 24.E2, 25.E2, 26.E2, 27.E2, 28.E2, 29.E2, 30.E2, 300.143, 284.791, 271.027, 258.609, 247.345, 237.076, 227.674, 219.029, 211.053, 203.669, 196.814, 190.432, 184.477, 1°, 306, -.1679, -.1669, -.1661, -.1654, -.1648, -.1644, -.1040, -.1636, -.1633, -.1630, -.1627, -.1624, -.1620, -.1616, .921, .919, .917, .915, .914, .912, .911,
           .911,
                                           .911.
                                                                          .911.
                                                                                                         .912.
```

For electron-ion interaction, the values for Q_{ei} , $A_{ei}^{(2)}$, and $A_{ei}^{(5)}$ are calculated as follows:

$$Q_{ei} = \frac{\pi}{2} \left(\frac{e^2}{4\pi\epsilon_0} \frac{1}{kT_e} \right)^2 \quad \text{in } \Lambda$$

$$\text{or } Q_{ei} = \frac{4.387 \cdot 10^{-10}}{Te^2} \quad \text{in } \Lambda$$

$$A_{ei}^{(2)} = -0.6 \\ A_{ei}^{(5)} = 1.3$$
(Ref. 4, for coulomb interaction)

3. Calculation of the Hall Parameter, β

According to the algorithm adapted [4] for the calculation of electrical properties of plasma, the Hall parameter, β , is

$$\beta = \chi \cdot \sigma \cdot B$$
,

where B is the magnetic field.

However, χ is also a function of the magnetic field, where [4]

$$x = (en_e)^{-1} \left(1 + [\Delta \tau_e^* v_t/(1 + \omega_e^2 \tau_e^{*2})]\right),$$

and

$$\omega_{e} = \frac{eB}{m_{e}}$$
.

The above equation for χ can be simplified, after substituting for the variables Δ , $\tau_e^{\ *}$, and ω_e , and the equation becomes:

$$x = (e n_e)^{-1} \frac{c_1 + B^2}{c_0 + B^2}$$

where

$$C_0 = (m_e/e)^2 (\tau_e^*)^{-2}$$
 and

$$C_1 = C_0 + \frac{5}{2} \left(\frac{m_e}{e} \right)^2 v_0^2$$
.

The influence of the magnetic field on the Hall coefficient, x, has been investigated for zero magnetic field, up to 20 T for a typical MHD plasma composition. The results are presented in Fig. II-1. These results indicate that there is a very weak dependence of the coefficient, x, on the magnetic field. This is a result of C_0 and C_1 being very small and, also, almost equal.

Based on these results, the effect of the magnetic field on the Hall coefficient, x, is neglected, and the equation for the computation of x is reduced to

$$x = (en_{\theta})^{-1}$$

4. Error Bounds on Electrical Conductivity

For the same method used to compute the electrical conductivity of the



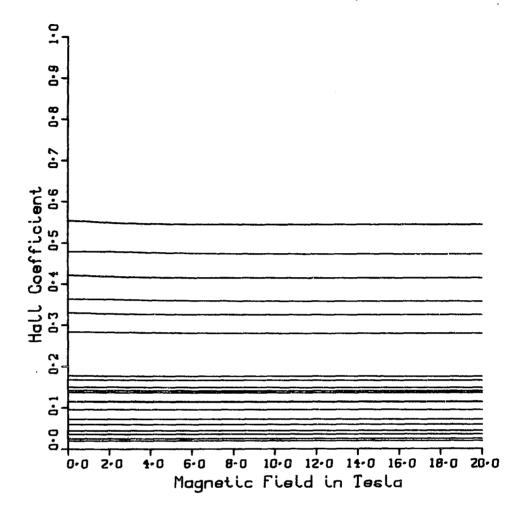


Fig. II-l Effect of the magnetic field on the Hall Coefficient (as defined in Ref. 4)

plasma, there are two major sources of errors that can contribute to the uncertainty of the computed values for conductivity. The first source of error is the thermochemical data for the heat of formation of some of the species, which can significantly affect the concentration of the electrons in the plasma. The most important species are KOH, Alo_2 , and Co_2 . The second source of error is the values for the momentum collision cross sections. Reference 7 discussed the error bounds on the electrical conductivity, which result from the uncertainty of the data of the first kind. At ANL, independent calculations were done to investigate such effects.

For CO₂: Using the JANAF data of December, 1975, σ is reduced by 5% at a reference point (p = 3.5 atm, T = 2600 K, for Montana Rosebud coal with 5% moisture, 85% slag rejection, and 34% oxygen enrichment) as compared to the case where CO₂ is neglected. Based on the investigation of References 7 through 9, it has been shown that, with the new data for the heat of formation for CO₂, there is very little effect of CO₂ on the electron concentration in the plasma and, correspondingly, very little effect on the electrical conductivity of the plasma. On this basis, until the JANAF tables are officially updated, CO₂ is neglected in our analysis.

For AlO₂: Using the old JANAF data of December, 1968, σ is reduced by approximately 13 %. The error bounds on the heat of formation in the new data (12/75) are smaller, which makes the range of uncertainty smaller (\sim 13 %) than before.

For KOH: Reference 7 showed that s can vary by as much as 50 % because of the uncertainty of + 3 kcal/mol in the heat of formation. Therefore, it is important to reduce experimentally the range of uncertainty. In the meantime, ANL uses the JANAF medium value.

C. Heat Losses

11/1

When run in the HP mode, the original NASA code yielded a single calculational point for each pressure input. The heat of formation of each reactant, as input or as computed, was multipled by the mole fraction of that reactant, and the products were summed to give the total enthalpy, HSUBO. Chemical equilibrium was then calculated for that total enthalpy

and each specified pressure. For many purposes, especially combustor studies, it is valuable to be able to find equilibrium properties as a function of combustion chamber heat loss. The new version has been adjusted to allow these studies.

These heat loss studies are performed by computing chamical equilibrium at an enthalpy that is reduced by some amount from HSUBO. Because enthalpy is a relative measure, a reference value, REFH, was added to the inputs. This is typically the enthalpy of the same reactants when brought to equilibrium at some standard set of conditions, and is treated as the enthalpy when heat loss is '100 %.' For a given fractional heat loss, f, the enthalpy at which equilibrum is computed is

H=HSUBO - f (HSUBO - REFH).

To save space, the heat loss fractions are input and stored in the T array, the same array as is used for temperatures in TP mode calculations. Because this array is internally counted by looking for the first zero element, and because no heat loss should be an option, the values actually input to T are from (1-f), rather than f.

Thus, for example, to compute the combustor temperature for a 2.5% combustor heat loss (based on the total thermal input), the HP input mode is used and the value 0.975 is entered in the T array of the NASA input.

D. Shorthand Data Input

The reactants for virtually all MHD combustor studies, and for a large number of other applications, conform to a standard model. A hydrocarbon fuel (coal, oil, natural gas, or some combination of them) of given chemical analysis and heating value is burned in a preheated air/oxygen mixture with some weight fraction of potassium, injected as K_2CO_3 . The air, fuel, and K_2CO_3 may be accompanied by some given amount of water.

Although, relatively straightforward, inputting this standard analysis data by REACTANTS cards is rather tedious. For use in coal combustion, the weight fractions must be adjusted from the 'as received' to the 'as determined' basis, in order to correct for various moisture content and ash rejection ratios. The weight fraction of air and pure oxygen must be adjusted to meet the desired 0_2 -enrichment and stoichiometric ratios. The weight fraction of K_2CO_3 must be adjusted to the proper fraction of potassium seeding. And a fictitious heat of formation for one reactant is required, so that enthalpy of the total mixture will correspond to the given heating value of the fuel as correctly adjusted to the 'as determined' basis.

To simplify treatment of this standard problem, a new input group, FUEL, has been added. As described in detail in Section IV, a FUEL group defines the following:

the chemical composition of the fuel,
fuel heating value,
moisture content,
ash rejection,
stoichiometry of the combustion,
O2-enrichment,
fuel temperature,
air temperature,
potassium (seed) weight fraction,
water-to-seed weight fraction.

Within the computational entry, NASA, the relative weights and the fictitious enthalpy of the carbon are computed. All the information from

the FUEL group is then stored in the appropriate internal arrays - duplicating the situation that would have occurred if the reduced data had been read from REACTANTS cards.

The computations that are performed in this setup process are as follows:

1) From the ultimate analysis of the fuel, the total fuel weight (TOT_WT) is readjusted to yield the specified value of fuel moisture content (FUEL.H2O ADJUST).

TOT WT = TOT WT*(1.0-H20 FRAC FUEL)/(1.0-FUEL.H20 ADJUST)

The 'as received' heating value for the fuel (FUEL.HHV) is readjusted to account for this moisture change.

HHV = HHV*(1.0-FUEL.H20 ADJUST)/(1.0-H20 FRAC FUEL)

2) From the specified fraction of ash rejection (FUEL.ASH_REJECT), the weights, weight fractions, and HHV are readjusted.

SLAG WT = total weight of ash content as received,

HHV = HHV*TOT_WT/(TOT_WT-ASH_REJECT*SLAG_WT),

TOT_WT = TOT_WT-ASH_REJECT*SLAG_WT,

 $X = X*(1.0-ASH_REJECT),$

where X is the weight of any ash species (e.g., SiO2, SO3).

3) From the input carbon (C), hydrogen (H), sulfur (S), and oxygen (O) content of the fuel and the stoichimetry (STOICH), the total weight of oxygen (OWT) needed in the oxidizer mixture is determined.

OWT = STOICH*(32.0*(C/12.0111) + 0.5*(H/2.016) + (S/32.066))-0, where the molecular weights of 0_2 , C, H_2 , and S are 32.0, 12.011, 2.016, and 32.066, in that order.

4) From the oxygen enrichment, the total amount of air (AIRWT) is determined.

AIRT = OWT/O2 AIR

5) From the air weight and oxygen weight, the weight of pure oxygen (O2WT) is determined, and, from the humidity (H2OA), the amount of water (H2OWTA) in the oxidizer mixture is determined.

02WT = OWT - 0.2315 * AIRWT

H2OWTA = H2OA*AIRWT,

where the mass ratio of oxygen in dry air is 0.2315.

6) From the air temperature (AIRT), its enthalpy (AIRH) is obtained from the polynomial fit.

7) From the seed fraction (SF), the fuel weight (FW), and the moisture fraction (H2OS), the amount of K_2CO_3 (SEEDWT) and the amount of water (H2OWTS) accompanying it are determined.

SEEDWT = SF*(FW+02WT+AIRWT+H20WTA)/(0.5658-SF*(1+H20S))

H2OWTS = H2OS*SEEDWT.

where 0.5658 is the mass ratio of K in K_2CO_3 .

8) From the heating value (HHV) of the fuel, its total weight, and its carbon, hydrogen, and sulfur content, the appropriate fictitious carbon heat of formation ('Hf') is determined.

Hf = -94052+12.011/C*(HHV*FW-H/2.016*6837.0-S/32.066*70947.0), where the heat of formation values for CO_2 , H_2O , and SO_2 are -94052, 68317, and 70947 cal/g, in that order.

These steps are applied to each FUEL group automatically, loading the internal arrays within the NASA code as if a set of REACTANTS cards had been furnished.

E. Thermodynamic and Electrical Properties Summary Output

When the output routine, NASAOUT, is called, it causes the printing of a summary sheet containing the overall thermodynamic properties of the equilibrium mixture. This summary consists of the following labeled columns of data.

<u>Column</u>	Label	<u>Description</u>
1	Р	pressure in atmospheres (input quantity)
2	Т	Temperature in kelvins (input quantity in TP mode)
3	MW	Equilibrium molecular weight
4	н	Enthalpy in joules/kg
5	S	Entropy in joules/kg·K (input quantity in SP mode)
6	RHO	Density in kg/m ³
7	СР	Specific heat in joules/kg·K
8	SON VEL	Sonic velocity in m/s
9	SIG	Electrical conductivity in S/m (present only if
		conductivity calculation required)
11	NE	Electron number density in $10^{20}/\mathrm{m}^3$ (present
		only if conductivity calculation is required)

These numbers are stored internally as the rows of the PL/1-based structure array, NASAP->OUT.A (cf Ref. 2) whenever the calculational entry, NASA, is called. Thus, for example, the values

NASAP->OUT.A(10,J)

for $J=1, 2, \ldots$, NASAP->IFIN refer, in order, to the conductivities resulting from each equilibrium point calculated. Because the pointer, NASAP, is an external variable, all of these data points are known to (and may be used in) the MAIN Program.

III. USER'S GUIDE

This section will describe in detail the input data to the NASA code and the output that may be obtained with the various options. The variable names into which the input data is stored are also listed and explained, because knowledge of these names is necessary to effect control from GPSAP over parameters in NASA.

A. Input Data

Data input to NASAIN, contained in file CARDIN, is divided into categories, each introduced by a keyword. The keywords and the type of input data following each are:

	Keyword	Type of data
1)	REACTANTS	Description of chemical reactants introduced into the combustor
2)	FUEL	Shorthand description of reactants
3)	NAMELIST	Type of problem, prescription of "givens," selection of options
4)	OMIT	Chemical species to be excluded in computing equilibria
5)	INSERT	Condensed species to be included in first iteration
6)	END	Signifies the completion of problem prescription

These keywords must appear, starting in Column 1. Any additional data on the same line (card) with REACTANTS, FUEL, NAMELIST, or END key cards is ignored; thus, comments may be entered in those lines.

REACTANTS, OMIT, and INSERT all trigger 'GET EDIT' statements, and the data must, therefore, occur in specified columns.

FUEL and NAMELIST trigger 'GET DATA' statements, allowing more freeform input. END returns program control to the calling (MAIN) program. Inclusion of this key card in the program vocabulary allows the other input categories to occur in any order. The various categories are discussed in detail below.

1. REACTANTS

Except for omitting the division of reactants into "fuels" and "oxidants," the REACTANTS group format has not been changed from that of the original NASA-Lewis code. Thus, the REACTANTS groups from any existing data sets for that code can be used directly in the present one. The names, meanings, and input format for each REACTANTS card (from 1 to NREAC, where NREAC is the total number of reactants read) are as follows:

Variable	Meaning	Format	Columns
NAME(N,I),NUM(N,I)	Atomic symbol and formula numbers for element I (maximum 5) of reactant N (if NAME(N,5) = '00,' the enthalpy is calculated by the code).	5(A(2),F(7,5))	1-45
PECWTS(N)	Relative weight or number of moles of reactant N in total reactants. Arbitrary base.	F(7,5)	46-52
MOLE	If MOLE='M' for any N, then variable MOLES is set to '1'b, signifying that all PECWTS are interpreted as relative numbers of moles. MOLES is alterable at execution time, MOLE is not.	A(1)	53
ENTH(N)	Assigned enthalpy in cal/g·mol [see note regarding NAME(N,5)].	F(9,5)	54-62
FAZ(N)	'S', if solid 'L', if liquid 'G', if gas	A(1)	63

RTEMP(N)	Reactant temperature in K	F(8,5)	64-71
DENS(N)	(optional) reactant density	F(8,5)	73-80
	in g/cm ³		

Note that column 72 is ignored. All of the above, except MOLE (but including MOLES), can be altered with 'AAA' statements at execution time, as shown in the examples in Section IV.

2. FUEL

The FUEL keyword is a local extension, designed to simplify data input for the most common MHD applications. After the keyword, any members of the structure FUEL are input as free-form assignment statements (e.g., FUEL.0 = 86.4) separated by commas and/or any number of blanks. The end of a FUEL group is designated by a semicolon. The names, meanings, and default values of the variables input in a FUEL group are

<u>Variable</u>	Meaning	Default
FUEL.C	Relative weight of carbon in fuel	0.0
FUEL.H	Relative weight of hydrogen in fuel	0.0
FUEL.S	Relative weight of sulfur in fuel	0.0
FUEL.N	Relative weight of nitrogen in fuel	0.0
FUEL.0	Relative weight of oxygen in fuel	0.0
FUEL.H20	Relative weight of H ₂ O in fuel	0.0
FUEL.Si02	Relative weight of SiO ₂ in fuel	0.0
FUEL.SO3	Relative weight of SO ₃ in fuel	0.0
FUEL.AL203	Relative weight of $A1_20_3$ in fuel	0.0
FUEL.CAO	Relative weight of CaO in fuel	0.0

FUEL.MGO	Relative weight of MgO in fuel	0.0
FUEL.FE203	Relative weight of Fe_2^{0} in fuel	0.0
FUEL.K20	Relative weight of K ₂ O in fuel	0.0
FUEL.TOT_WT	Total weight to which the above components are normalized	100.0
FUEL .T	Temperature of fuel	300.0
FUEL.HHV	Higher heating value of fuel in cal/g	0.0
FUEL .DENS	density of fuel	0.0
FUEL.STOICH	Ratio of total oxygen supplied in fuel, air, and pure 0, to total oxygen required for complete combustion of all carbon, hydrogen, and sulfur	1.0
FUEL.02_AIR	Weight ratio of the total 0_2 to the air + 0_2 mixture supplied	0.2315
FUEL·H2O_AIR	Weight of water carried by the air, as a fraction of the dry air weight	0.0
FUEL.AIR_TEMP	Temperature of air and its water	300.0
FUEL.02_TEMP	Temperature of the pure 0 ₂	300.0
FUEL.SEED_FRAC	Weight fraction of potassium in total to be introduced as ${\rm K_2CO_3}$	0.01
FUEL.H2O_SEED	Weight of water accompanying the K_2CO_3 as a fraction of the K_2CO_3 weight	0.0
FUEL.H20_ADJUST	Weight fraction of H ₂ O required in the fuel. Thus, if FUEL.H ₂ O ADJUST is not equal to the moisture fraction in the fuel, the weight fractions of water and the HHV of the fuel are readjusted to this new moisture fraction.	0 . 0
FUEL.ASH_REJECT	Fraction of the total ash content of the fuel to be rejected	0.0

Notes:

(1) If FUEL.C is not entered, the entire FUEL group is ignored.

- (2) The seed fraction is calculated as a fraction of the total of the reactants generated by this FUEL group.
- (3) As many as three FUEL groups and one REACTANTS group may be entered to describe the reactants for a given problem. At input, each FUEL group is tagged by the keyword, FUEL, and input as FUEL.xxx=vvv
- (4) For modification at execution time, members of the second and third FUEL groups entered are modified by

AAA 'FUEL2.xxx' vvv

and

AAA 'FUEL3.xxx' vvv,

respectively.

3. NAMELIST

The keyword, NAMELIST, triggers a GET DATA statement, which reads a number of parameters and switches that control the calculations to be performed. Although the basic idea is the same as in the original NASA-Lewis code, considerable modification of this input categroy was necessary, both to accommodate the PL/I environment and to serve the locally available options. The variables, types, meanings, and defaults are, at present:

Variable	Dimension	Type	<u>Meaning</u>	<u>Default</u>
TP	1	BIT(1)	If 'l'B, equilibria calculated at given temperatures and pressures	'0 'B
НР	1	BIT(1)	If 'l'B, equilibria calculated at given enthalpies and pressures	'0'B
SP	1+-	BIT(1)	If 'l'B, equilibria calculated at given entropies and pressures	'0'B
0T <u>1</u> 0	1	BIT(1)	If 'O'B, JANAF chemical table search is skipped whenever the preceding NASA call had the same elements and omissions	'0'B

<u>Variable</u>	<u>Dimension</u>	<u>Type</u>	Meaning	<u>Default</u>
MMHG	1	BIT(1)	If 'l'B, pressures are in mm Hg	'0'B
INGH	1	BIT(1)	If 'l'B, pressures are in in. Hg	'0'B
PSIA	1 .	BIT(1)	If 'l'B, pressures are in psi	'0'B
NSQM	1	BIT(1)	If '1'B, pressures are in N/m ² . Note that, if none of MMHG, INHG, PSIA, or NSQM, then pressures are in atmospheres	'0'B
FIT	1	BIT(1)	If 'l'B, summary output is written onto file FITDAT for subsequent polynomial fit	'0'B
CONDUC	1	BIT(1)	If 'l'B, conductivity routine is called after equilibrium calculation	'0'B
IONS	1	BIT(1)	If 'l'B, ionized species are included in equilibrium calculation	'l'B
PRINT	1	BIN(15)	Determines the amount of inter- mediate output	0
IDEBUG	1	BVIN(31)	<pre>If > 0, causes equilibrium iterations to be output</pre>	0
TRACE	1	DEC(6)	If > 0, signifies that, when mole fractions are printed, all species with fraction > TRACE are shown	0
REFH	1	DEC(6)	Reference enthalpy: enthalpy of the same reactants if brought to equilibrium at standard temp and pres. In TP mode, this is used as a "zero set" for the reported enthalpies. In HP mode, it is used as a reference point to define heat loss	0
S0	1	DEC(6)	Total entropy at which equilibria are calculated in SP mode	0
Р	26	DEC(6)	Schedule of pressures at which equilibria are to be calculated. End of the schedule is signified by the first 0	26*0

Variable <u>Dimension</u> <u>Type</u>		Туре	pe <u>Meaning</u>	
T	26	DEC(6)	If TP mode, schedule of temperatures (in K) at which equilibria are calculated. If HP mode, schedule of heat retention fractions at which equilibria are calculated (see below)	26*0

The use of the T array in HP mode is a locally written extension to the original NASA-Lewis code. In this application, the values l-T(I) represent the combustor heat loss as a fraction of the enthalpy difference between the reactants as introduced to the combustor and the enthalpy REFH. Thus, e.g., if REFH is the enthalpy of the same reactants at 300 K, l atm, then a value l = .96 represents a heat loss of 4% of the total thermal input.

Note that, in a NAMELIST input from a call to NASAIN, all variables are redefined to their default values before the NAMELIST is read. This can be important when GPSAP is used to do parameter sweeps and optimizations.

4. OMIT and INSERT

OMIT and INSERT cards are identical in both function and format to those in the original NASA-Lewis code. The keyword is followed by the names (chemical symbols) of from one to four species, starting in columns 16, 31, 46, and 61. The names on OMIT cards signify particular species that are to be deleted from consideration in computing the equilibrium composition. INSERT cards contain the names of condensed species that are assumed to be present when the program is began. The variables, NSERT and NOMIT, contain, respectively, the number of inserted and omitted species; ENSERT (3,30) and OMIT(3,3) [both CHAR(4)] contain the species names. At execution time, the last species in either list may be removed by decreasing NSERT or NOMIT. A species may be changed to a new one by assigning the appropriate new character strings to the array elements. The name of the Nth species is left-justified in the 12-character field,

(XXX(1,N), XXX(2,N), XXX(3,N)),

where XXX stands for either OMIT or ENSERT. If the old species name is longer than the new one, the extra spaces must be explicitly blanked. A new species may be added to the list by increasing NSERT or NOMIT and assigning the new name to the array.

5.0 END

The END card does not signal additional input but, rather, indicates that a problem is now fully specified and that control should be returned to the main program for further processing.

B. Program Output

Primary output from the calculation is the thermodynamic and electrical properties summary printed upon a call to NASAOUT. This output consists of a table for all the points on the P and T schedules of the following:

```
pressures (atm)

temperatures (K)

molecular weights (kg/kg-mol)

enthalpies(J/kg)

entropies (J/kg·K)

densities (kg/m³)

specific heats (J/kg·K)

gas constants

sonic velocities (m/s)

conductivities (S/m)

electron number densities (10<sup>20</sup>/m³)
```

These same values are also accessible from the driver as NASP->OUT(I,J), where I goes from 1 to 13 and refers to the properties (in the order given)

and J goes from 1 to the number of output points (T varies more rapidly than P).

In addition, various amounts of intermediate output may be obtained, depending on the value of PRINT.

If PRINT=0, all that is obtained is a copy of the input data and any error messages. Error messages will be on file FT06F001 if errors occurred in the equilibrium calculation and on SYSPRINT otherwise. The messages and their meanings are identical to those in the original code and will not be repeated here.

If PRINT=1, the output also includes a list of the species considered each time the JANAF thermodynamic data are scanned, the value (HSUBO) of the total enthalpy of the reactants as introduced, the number of gram atoms of each element present per kg of the mixture (BO's), a listing of any inserted species, and a list of the REACTANTS cards that are equivalent to any FUEL groups used and a list of the reactants, weight fractions (normalized to a total of 1), enthalpies (as input or calculated), phases, reference temperatures, and densities (if input), as well as the net mixture density at input.

PRINT=2 adds a brief summary of the net thermal and electrical properties printed once every 13 points on the calculation schedule.

PRINT=3 yields all of the above plus 1) a listing of the mole fractions for every species of mole fraction greater than or equal to 5 \times 10⁻⁶ if TRACE=0 (TRACE otherwise), 2) a trace of the LaGrange multiplier values, and 3) the names of added or removed condensed species as the equilibrium calculation proceeds (this last is on file FT06F001).

IV. ILLUSTRATIVE EXAMPLE

Two examples will both demonstrate the method of using the new GPSAP-driven NASA code and illustrate the new capabilities it affords. Any confusion as to variable names and meanings may be resolved by reference to Section III.

A. Example 1

1. Problem

Determine the thermodynamic and electrical propeties of a coal burned in an air/oxygen mixture. The coal has an 'as received' composition analysis of:

<u>Species</u>	<u>Weight</u>
С	52.2
Н	3.38
S	0.8
N	0.8
0	11.02
SiO ₂	1.732
S0 ₃	2.055
A1203	1.214
CaO	1.481
Mg0	0.517
Fe ₂ 0 ₃	0.499
H ₂ 0	24.30

Before being burned, the coal is dryed to 5% moisture content; furthermore, it is assumed that the ash rejection is 75%. (For the analysis, it

is assumed sufficient to remove 75% of the ash species before the combustion, rather than 75% of the condensed species during the combustion process.)

The oxygen/air mixture is to have a preheat temperature of 1000 K and 30 mass % 02. Sufficient air will be added to obtain a stoichiometric ratio of 0.92. The air is assumed to have moisture content of 1 mass %.

The combustion process is also seeded with 1 mass % potassium. The seed and coal are input at a temperature of 300 K. Also, the higher heating value of the coal is assumed to be 4968 cal/g.

Specifically, the flame temperatures of the combustion are to be determined for pressures is of 4, 5, 6, 7, and 8 atm with no heat loss. In addition, the thermodynamic and electrical properties are to be determined for pressures in the range of 1-8 atm, in 1 atm increments, and for temperatures in the range of 1600-3000 K, in 200 K increments.

2. Input

The total input required for this analysis is shown in Fig. IV-1. This input will now be discussed in detail, on the basis of the line numbers shown at the left in the figure.

Lines 1 to 29 comprise the standard JCL, common to all GPSAP analysis. The in-line procedure, SYSTEM, that is executed (line 29) is composed of three steps: ONE, PLO, and TWO. Step ONE executes GPSAP, which reads the structure in file STRUCIN and expands it into a PL/1 MAIN program on scratch file SYSDRV. Step PLO compiles that program, and step TWO loads it, using library SYSLTB.

Lines 33 to 38 complete the JCL, defining the various data files referenced by NASA. CNDCT contains the table of coefficients used to compute the conductivity. JANAF contains the thermochemical data (JANAF table) used by NASA. FITDAT is a scratch file used to store the NASA

Figure IV-1

```
1.
     JOB
     CUA
 2.
 3.
     //SYSTEM PROC
     //ONE EXEC PGM=GPSAP
     //ST EPLIB
                 DD
                      DSN=C115.B25236.SYSLIB.LOAD.DISP=SHR
                 ממ
                      DDNAME=STRUCIN
     //STRUCT
     //SYSPRINT DD
                      SYSOUT=A,DCB=(RECFM=VBA,LRECL=137,BLKSIZE=1511)
 7.
     //SYSDRV
                      DISP= (NEW, PASS), UNIT=SASCR, SPACE= (TRK, (2,1))
 8.
                 DD
     //PLO EXEC PGM=IFLOAA, PARM='NS, NA, NX, NAG, NOESD, NSTG
 9.
     //STEPLIB DD
10.
                     DSN=PLI.OPT.LINKLIB.DISP=SHR
     //SYSIN
                 ממ
                      DSN=*. ONE. SYSDRY, DISP= (OID, DELFTE)
11.
                      UNIT=SASCR, SPACE= (CYL, 4) ,
12.
     //SYSLIN
                 DD
13.
         DISP = (NEW, PASS), DCB= (RECFM=PB, LRECL=80, BLKSIZE=3120)
     //SYSPRINT DD DUMMY
14.
     //SYSPUNCH DD
15.
                      DUMMY
     //SYSUT1
                 DD
                      SPACE= (CYL, 4) , UNIT= (SASCR)
16.
17.
     //TWO EXEC PGM=LOADER, REGION= 150 K, COND= (9, LT, PLO)
     //SYSLIB
18.
                 DD DSN=SYS1.PLIBASE, DISP=SHR
19.
                 DD
                      DSN=SYS1.FORTLIB, DISP=SHP
20.
     11
                 DD
                      DSN=C115.B25236.LOAD,DISP=SHR
                      DSN=*.PLO.SYSLIN, DISP=(OLD, DELETE)
     //SYSLIN
21.
                 DD
22.
     //SYSLOUT
                 DD
                      SYSOUT=A, DCB= (RECFM=FB, LPECL=121, BLKSIZE=1573)
23.
     //SY SPN CH
                 DD
                      DUMMY
24.
     //SYSPRINT DD
                      SYSOUT= A, DCB= (RECFM=VBA, IRECL=137, BLKSIZE=1511)
     //SYSPUNCH DD
25.
                      DUMMY
                      SYSOUT=A, DCB= (RECFM=FB, LRFCL=133, BLKSIZE=1596)
26.
     //PROGREP DD
27.
     //FT06F001 DD
                      SYSOUT=A, DCB= (RECFM=FB, LRECL=133, BLKSIZE=1596)
     // PEND
28.
     // EXEC SYSTEM
29.
     //ONE.STRUCIN DD *
30.
31.
      MMM NASAIN NASA NASAOUT
32.
      MMM NASAIN NASA NASAOUT
33.
     //TWO.CNDCT DD DS N= B25445. TABLE. DATA, DISP=SHR
34.
     //JANAF
                   DD DSN=B25445.THERMO.DATA,DISP=SHR
35.
     //SUMMARY
                   DD
                        SYSOUT= A, DCB= (RECFM=FB, LRECL=133, BLKSIZE=1596)
                   DD DISP= (NEW, PASS), UNIT=SASCR, SPACE= (TRK, (5, 5)),
36
     //FITDAT
37
          DCB=(RECFM=FP, LPECL=80, BLKSIZE=1520)
     //CARDIN
38
                   DD
39
     FUEL
40
       FUEL.C=52.2 FUFL.H=3.38 FUEL.S=0.8 FUEL.N=0.8 PUEL.O=11.02
41
       FUEL.SIO2=1.732 FUEL.SO3=2.055 FUEL.AL203=1.214 FUEL.K20=0.0
42
       FUEL.CAO=1.481 FUEL.MGO=0.517 FUEL.FE203=0.499 FUEL.H20=24.3
       FUEL.ASH_REJECT=0.75 FUEL.H2O_ADJUST=0.05 FUEL.SEED_FRAC=0.01 FUEL.H2O_SEED=0.00
43
44
45
       FUEL.HHV=4968. FUEL.STOICH=0.92
46
       FUEL.T=300. FUEL.AIR_TEMP=1000. FUEL.O2_TEMP=1000.
47
       FUEL.H20_AIR=0.01 FUEL.02_AIR=0.3 ;
48
49
     NAMELISTS
50
       HP=^{1}B FIT=^{0}B P(1)=8 P(2)=7 P(3)=6 P(4)=5 P(5)=4
51
     END
52
     NAMEL ISTS
53
       TP= '1'B CONDUC= '1'B FIT= '1'B PRINT=2
       P(1) = 8 P(2) = 7 P(3) = 6 P(4) = 5 P(5) = 4 P(6) = 3 P(7) = 2 P(8) = 1 P(9) = 0.5
54
       T(1) = 3000 T(2) = 2800 T(3) = 2600 T(4) = 2400 T(5) = 2200 T(6) = 2000
55
56
       T(7) = 1800 T(8) = 1600 T(9) = 1400 T(10) = 1200 T(11) = 1000 T(12) = 800
57
58
     BND
```

answers for further processing by other procedures. The input data to NASAIN is found in file CARDIN. The names, CNDCT, JANAF, FITDAT, and CARDIN, are built into the NASA code and, hence, cannot be changed.

This problem requires two distinct operations, 1) the determination of the flame temperatures, and 2) the calculations of the thermodynamic and electrical properties. Each of these, in turn, requires three processes:

1) inputting data, 2) calculating the results, and 3) printing the results. Line 31 indicates the simplest GPSAP input for performing these three tasks with the NASA code. The MMM statement will cal? the NASAIN entry, which will read the input data on file CARDIN down to the first END keyword. It will then call NASA and perform the computations as dictated in the input data. Finally, it will call NASAOUT to print a summary of the results, which will be the flame temperatures in this case.

The second line, 32, in file STRUCIN will perform exactly the same. This time, the first call to NASAIN will read down to the next END keyword, encompassing the inputs necessary to perform the calculations of the thermodynamic properties.

Lines 39 to 58 represent the input data to the NASA code for this problem. As stated, the problem can be input directly without any prior anlaysis. Lines 39 to 47 represent the FUEL group, giving the coal composition, moisture readjustments, ash rejection corrections, stoichiometry, etc. Lines 52 to 57 represent the NAMELIST input. Thus, the HP mode will be used with the desired pressures. The END keyword in line 51 indicates the end of the data that will be read-in in this call to NASAIN.

Lines 52 to 56 indicate the NAMELIST input necessary to carry out the thermodynamic properties calculations. Line 58 signifies the end for the second NASAIN call.

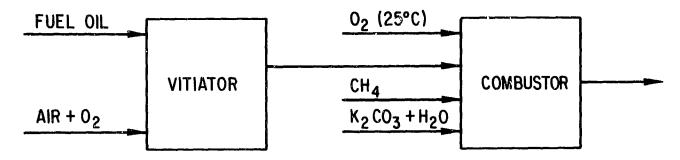


Fig. IV-2 Schematic diagram of proposed combination combustor

B. Example 2

1. Problem

Operate the combination combustor, depicted in Fig. IV-2, of a proposed experimental MHD facility. The vitiator burns a mixture of No. 2 fuel oil and preheated oxygen-enriched air. Oxygen is added to the air to establish 30 mass % 0_2 . In the combustor, the output of the vitiator is combined with additional, room-temperature oxygen, with methane, and with a one-to-one mixture of potassium carbonate and water. Potassium is to form 1 mass % of the total reactant mixture.

The composition of No. 2 fuel oil will be taken as follows:

C 86.4%

H 12.7%

S 0.7%, and

O 0.2%,

with a higher heating value of 10,877.3 cal/kg.

For a number of different air preheat temperatures and a number of different (total) oxygen enrichment levels, it is desirable to know the relative amounts of oil and methane necessary to achieve stated flame temperatures at 3 atm pressure, 7% system heat loss. Specifically, for

T preheat = 700, 800, 900, 1000 K, and 02 enrichment = 0.30, 0.35, 0.40, we wish to obtain,

T flame = 2900, 3000, 3100 K.

It is required, further, that these temperatures be achieved with the minimum possible amount of oil, i.e., an optimal stoichiometry will be found simultaneously.

2. Analysis

As will be seen, the availability of GPSAP code-creation and optimization capabilities allows this entire study to be performed in a single computer job. A small amount of preliminary analysis is necessary, however.

Note, first, that only the final state of the combustion products is of interest. This final state is unaffected by the fact that combustion is carried out in a two-stage process. We will, accordingly, merely compute the equilibrium state that exists when all reactants are combined. Second, note that the thermodynamic properties under investigation are intrinsic quantities, unaffected by the total reactant mass. Therefore, all reactants will be scaled to 100 grams of methane.

The heating value for methane is not stated and, presumably, is not known. Conversely, the heating value for fuel oil is given, and we wish to avoid calculating the corresponding 'Hf.' As now composed, subroutine SETUP (which translates the FUEL groups into NASA input arrays) is rather awkward for this sort of FUEL/REACTANTS mixture. Therefore, it is easiest to input only the fuel oil through the FUEL group, and to adjust the amounts of air, 0_2 (hot and cold), K_2C0_3 , and H_2O explicitly from the input structure.

The question that must be answered is, given,

CH = 100 grams,

oil = X grams,

stoichiometry ratio = ST,

02 enrichment = 02 ENRICH,

T-preheat = AIRT,

seed fraction = 0.01, and

 $K_2CO_3/H_2O = 1.0$,

find the weight of K_2CO_3 , H_2O , O_2 (hot), air, and O_2 (cold). Once stated the problem is answered fairly easily. The total weight of oxygen necessary (air and pure oxygen) is

02TOT = ST *
$$\left\{ \begin{bmatrix} 2* & \frac{100}{16.043} + \frac{0.864*X}{12.011} + \frac{1}{2} & \frac{0.127*X}{2.016} \\ & + \frac{0.007*X}{32.066} \end{bmatrix} * 32 - 0.002*X \right\}$$
= ST * $\left[398.928 + 3.315 * X \right]$.

where, in the first entry, the four terms within the square brackets are, in order, the moles of methane and the moles of C, H, and S in the oil; the last term is the mass of oxygen already found in the fuel oil. From this, the weight of the air is

AIRW = 02TOT/02 ENRICH.

The hot 0_2 -plus-air is to contain 30 mass % 0_2 ; therefore the weight of hot 0_2 is

02 HOT = 0.3 *AIRW,

and the weight of cold 0_2 is

02 COLD = 02TOT - 02 HOT.

Finally, the weight of potassium carbonate and of its water of hydration are determined by the requirement that potassium, which constitutes 0.5658 of the weight of K_2CO_3 , is to comprise 0.01 of the total weight. The seed weight and the weight of H_2O are, thus, each

SW = 0.01*(100 + X + AIRW + 02 HOT + 02 COLD)/0.5658.

These manipulations are, of course, slightly tedious. It should be noted, however, that this is all that is required for the entire study with the present form of the code. Under the old regime also, all the same steps were required. In addition, 'Hf' would have to be calculated. And, finally, all formulae would have to be evaluated for each set of the X, ST, and O_2 _ENRICH to be studied.

3. Input

Only the input structure on file STRUCIN and the input data to NASAIN on file CARDIN will be explained. The rest of the JCL is similar to that for Example 1. Again, the input, shown in Figure IV-3, will be discussed on the basis of the line numbers shown on the left.

The REACTANTS cards, introduced by line 52 and contained on lines 53 to 58, are the same as those in the previous version of the code (except that separation into 'fuels' and 'oxidants' has been eliminated). The ${\rm CH_4}$, ${\rm K_2CO_3}$, ${\rm H_2O}$, preheated ${\rm O_2}$, air, and cold ${\rm O_2}$ are described in turn. These will be known to NASA as reactants 1 through 6. The relative weights of all but reactant 1 (${\rm CH_4}$), as well as the enthalpy and reference temperature of reactant 5 and the reference temperature of reactant 4, are dummies, because they will be changed by the MAIN Program (these numbers could, in fact, be deleted). The blank line (126) is necessary to indicate the end of the REACTANTS group.

The FUEL group, lines 60 through 63, defines the fuel oil and the relative weights of carbon, hydrogen, and oxygen. Because STOICH and SEED FRAC default (see Section III) to 1.0 and 0.01, respectively, it is

Figure IV-3

```
//ONE.STRUCIN DD *
      MMM WASAIN ;
BBB A AIRT = 700 800 900 1000
 3
         AIRH=-2059.76+AIRT* (6.713+
 4
 5
                            BIRT# (2.3485E-4 +
 6
7
                             AIRT* (3.823E-7 -
                              AIRT *1.174F-10)));
         BBB B 02_ENFICH = 0.3 0.35 0.4;

BBB C T0 = 2900 3000 3100;

BBB D X 100 10 500

ST 0.9 0.75 1.15;

02T0T=ST*(398.928 + 3.315*X);
 8
10
11
12
                   AIRW=02_ENRICH *02TOT;
13
14
                   02_HOT=0.3*AIPW;
                   O2_COID=MAX(0.001, O2TOT-O2_HOT);
SW= (100+X+AIRW+O2_HOT+O2_COLD)/54.58;
15
16
17
                AAA 'FUEL. TOT_WT'
                                        X
                      'PECWTS (2)
18
                                        SW
                      'PECWTS (3) '
19
                                        SW
20
                      PFCWTS (4)
                                        O2_HOT
21
                      * RTFMP (4) *
                                        AIRT
22
                      PRCWTS (5)
                                        AIRW
23
                      'ENTH(5)'
                                        AIRH
                      * R TEMP (5) *
                                        AIRT
25
                      PECUTS (6)
                                        02_COLD
                                        40 B
26
                      · HP ·
27
                      * TP *
                                        11'B
                      FIT.
                                        * 0 * B
28
29
                      'CONDUC'
                                        .0.B
30
                                        0
                      'PRINT'
                      *REFH*
                                        0.0
31
32
                      'P(1)'
                                        1.0
33
                      'T(1)'
                                        300.0
34
35
                MMM NASA ;
                      · HP
                                       111B
36
                AAA
37
                      · TP
                                       *0 * B
38
                      'P (1) '
                                       3.0
                      17 (1) 1
39
                                       0.93
40
                      * REFH*
                                       NASAP->A(4, NASAP->IFIN) ;
               MMM NASA
41
                    D NASAP->A(2,1) = T0;
D X HAXIT=150 ACC=1.0E-2 DEL=1.0E-3
42
               CCC
43
44
                PUT SKIP EDIT ("AIRT=", AIRT, "ONO="ONO, "TO=", TO,
45
                    'OIL WT=',X,'STOICH=',ST)
46
                    (5 (A,E(12,5),X(5)));
               HMM NASACUT ;
47
LA
             EEE C
49
         EEE B
                   ;
      EEE A
50
     //CARDIN
51
                  DD
52
     REACTANTS
53
    C 1.00
              H 4.00
                                                 00
                                                             1.5000
                                                                                G 298.1
               C 1.00
O 1.00
54
    K 2.00
                           0 3.00
                                                             18.902
                                                                                S 298.15 P
55
    H 2.00
                                                 CO
                                                             18.902
                                                                                L 298.15 F
     0 2.00
                                                 00
                                                             238.177
                                                                                G 1073.5 0
                                                             693.489 5730.34 G 1073.0 O
57
    N 1.5618 O 0.4196 ARO.0093 C 0.0003
58
    0 2.00
                                                 00
                                                             0.001
                                                                                G 298.15 O
59
60
     FUEL
       FUEL.C=86.4 FUEL.H=12.7 FUEL.S=0.7 FUEL.O=0.2
61
       FUEL.STOICH=0.0 FUEL.SEED FRAC=0.0 FUEL.T=298.15
62
       FUEL.HHV=10877.3 ;
63
64
65
     NAMELISTS
       PRINT=1 ;
66
67
68
     END
```

necessary to zero them explicitly to avoid introducing additional air, oxygen, and seed. Input, here, is terminated by the semicolon; the blank line (64) is cosmetic only.

Nearly all of the information input in the NAMELISTS group will be controlled by the MAIN Program. An exceedingly abbreviated NAMELISTS is included to set the defaults and to establish the desired amount of output on the 'production' runs. The END card returns control to the MAIN program.

The MAIN Program, which will perform the desired study, is described by the GPSAP input structure, lines 2 to 50. Line 31 calls the input module, NASAIN, which reads file CARDIN. Note that the reactants cards are read directly into arrays NAME, NUM, PECWTS, ENTH, FAZ, and RTEMP. The information from the FUEL group, on the other hand, is not translated into these arrays at that time.

The three BBB loops— lines 3 to 50, 8 to 49, and 9 to 48—sweep the desired parameter space in air temperature, oxygen enrichment, and flame temperature. Lines 4 to 7 calculate the air enthalpy for the specified temperature. The optimization loop (lines 10 to 43), carried out for each (AIRT,02ENRICH,TO) combination, is actually the heart of the computations. The BBB statement in lines 10 and 11 states that an optimization study is to be performed and that the independent variables are X (the weight of oil) and ST (the overall stoichiometry). The starting value of X is 100 (one-to-one CH and oil) and will be varied automatically between 10 and 500; ST begins at 0.9 and remains in the interval 0.75< ST < 1.15. For given values of X and ST, the amounts of air, hot oxygen, cold oxygen, seed, and water of hydration are calculated in lines 12 to 16.

The AAA statement, lines 17 to 34, alters a large number of parameters within the next calculational entry called--here, the NASA call in line 35.

The total weight of the fuel group (the oil) is adjusted to X in line 17. The seed weight, SW, is assigned to the potassium carbonate (line 18) and to the water (line 19). Reactant 4, the preheated oxygen, is assigned weight 02_HOT and temperature AIRT; the air (reactant 5) is given weight AIRW, enthalpy AIRH, and temperature AIRT, and the cold oxygen weight, 02_COLD, is assigned to reactant 6. Note that a reactant may not have negative or zero relative weight; accordingly, 02_COLD has been adjusted to remain > 0.001.

Lines 26 to 34 define a typical reference enthalpy run. TP is true ('1'B), because the enthalpy is to be derived at given pressure (1 atm) and temperature (300 K); FIT and CONDUC are false, because polynomial fitting is not required at this point and conductivity is not needed. REFH is zeroed, so that the derived enthalpy is not distorted in successive iterations by the reference value derived in the preceding iteration. The pressure P(1) is assigned to 1 atm, and the temperature T(1) to 300 K.

After the reference run in line 35, the AAA statement in lines 36 to 40 establishes an HP run at 3 atm and 7% heat loss. The value of the enthalpy at the last calculational point from the previous NASA call (i.e., at P = 1 atm, T = 300 K) is assigned to REFH.

In line 41, NASA is called to calculate the flame temperature at 3 atm, 7% heat loss. Line 42 states that the optimization in this BBB loop will be carried out under the constraint that this flame temperature be the given value TO. The EEE statement in line 43 identifies the end of the optimization loop and states that X and ST are to be varied so as to minimize X subject to the above-stated constraint on flame temperature. MAXIT identifies the maximum number of times that lines 10 to 43 may be

executed (for each value of AIRT, 02_ENRICH , and TO) in attempting to perform the optimization. DEL is the relative step size the optimizer uses for computing derivatives, and ACC is the minimum accuracy to which the constraint and optimization are to proceed. These last two are decreased from their defaults (10^{-6} and 10^{-4}), both to save computation time and to avoid numerical difficulties associated with the fact that NASA is a single precision routine.

Line 47 then calls the NASAOUT entry to print the thermodynamic properties table, which will consist, in this case, of only one line, representing the properties at the last call to NASA.

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APPENDIX

NAMELISTS

HP='1'B FIT='0'B P(1)=8.0 P(2)=7.0 P(3)=6.0 P(4)=5.0 P(5)=4.0;

ALH ALO- ALO- ALO- ALCO- ALCO- ALCO2+ C(S) C(S) CCN COC C20C CA(L) CAOCH2(S) FEO3H2(S) FEO3H2(S) FEO3H2(S) FEO4H2 HCO HNO3 HCO HNO3 HCO NO+ NCC NO+ NCC NO+	M3 CH+ S(S) SN SS SIC SIN SIOS(L)
2 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	12/70 11/2/70 12/65 12/65 12/64 12/67 12/67
AL+ ALO+ ALO2 ALCO2 ALCO2 ALCO2 ARCO2 ARCO2 CRC CRC CRC CRC CRC CRC CRC CRC CRC CR	N205 OH O3 SH SS SI+ SIC(S) SICC(S)
22222222222222222222222222222222222222	112/64 1 5/70 1 5/70 1 12/65 1 12/65 1 5/67 1 5/67
AL ALO ALO ALO ALCO ALCO ALCO ALCO AR CC CC CC CC CCAC CCAC CCAC CCAC CC	MCO4 0- CC2- SO3 SO3 SIH+ SIO2(S) SIZ
20111111111111111111111111111111111111	46/6/11 46/6/11 46/6/11 46/6/11 46/6/11
AL(L) ALN ALOH+ ALOH+ AL20 AL20 AL203(L) C+ C+ COZ CCR(S) FE(S) FE(S) FE(S) FE(S) FE(S) KCR(S)	N20+ 0+ 02 5 S SO2 ST(L) SIH SIS
26.25.25.25.25.25.25.25.25.25.25.25.25.25.	12/75 12/66 12/65 12/65 12/65 12/66 17/25 17/25
9	J12/64 N20 J6/74 O J12/70 OH- J12/65 S(L) J 6/71 SO J 3/67 SIC2 J 9/67 SIC2 J 9/67 SIC2 J 3/67 SIC2 J 3/67 SIC2
SPEC	

SIG	
SON VEL	9.3167E+02 9.3028E+02 9.265E+02 9.2673E+02 9.2673E+02
БАМ	1.1542E+00 1.1531E+00 1.1519E+00 1.1504E+00 1.1485E+00
ង	2.8262E+03 2.8555E+03 2.9116E+03 2.9570E+03 3.0362E+03
RHO	1.0779E+00 9.4503E-01 8.1203E-01 6.7262E-01 5.4434E-01
w	8.9958E+03 9.0331E+03 9.0761E+03 9.1271E+03 9.1595E+03
I	2.9776E+01 -2.6277E+04 2.9753E+01 -2.6276E+04 2.9737E+01 -2.6275E+04 2.9712E+01 -2.6277E+04 2.9681E+01 -2.6277E+04
¥	2.9753E+01 2.9753E+01 2.9757E+01 2.9712E+01 2.9681E+01
) —	2.6861E+03 2.6861E+03 2.6778E+03 2.6679E+03 2.6579E+03
۵	8.0000E+00 7.0000E+00 6.0000E+00 5.0000E+00 4.0000E+00

THERMODYNAMIC PROPERTIES SUMMARY

HSUB0 =-3.16014290E+00 (KG-MOL)(DEG K)/KG

KG-ATCMS/KG

NAMEL ISTS

TP='1'B,CONDUC='1'B,FIT='1'B,LONG=2 P(1)=8, P(2)=7, P(3)=6, P(4)=5, P(5)=4, P(6)=3, P(7)=2, P(3)=1 P(9)=.5 T(1)=5000 T(2)=2800 T(3)=2600 T(4)=2400 T(5)=2200 T(6)=2000 T(7)=1800 T(8)=1600 T(9)=1400 T(10)=1200 T(11)=1300 T(12)=860;

Ş

HSUB0 =-3.16014290E+00 (KG-MOL)(DEG K)/KG

80	.39486629E-0	. 19954345E-0	4.16722872E-05	.30382562E-0	.73517959E-0	.20364417E-0	.94325738E-0	0272354E-0	.60952735E-0	.55750519E-0	.56555335E-0	.000000000.
KG-ATOMS/KG	U	æ	ഗ	z	0	IS	AL	ÇÞ	37	¥	AR	w

to the contract of the second second

STATE TEMP D	. 087186 -9445.063 \$ 300.00 0.0000 . 005545 13.478 6 300.00 0.0000 . 001536 14.745 \$ 300.00 0.0000 . 001536 12.814 6 300.00 0.0000 . 018406 12.892 6 300.00 0.0000 . 018406 -68254.125 L 300.00 0.0000 . 010723 -217679 188 \$ 300.00 0.0000	-151768.500 \$ 300.00 -157203.750 \$ 300.00 -274845.188 \$ 300.00 5426.994 6 1000.00 5152.984 6 1000.00
HT FRAC	0.057 0.001 0.001 0.001 0.000	C 0.00030 0.000
CHEMICAL FORMULA	C 1.00000 S 1.00000 N 2.00000 D 2.00000 H 2.00000 O 1.00000 SI 1.00000 O 2.00000	CA 1.00000 0 1.00000

7.0000	3000.0	. 1953-4	247.1	2.2479	28.820	1.01779	1.3789	0.9235	1.1418	994.1	.6511 1	.2233-1	.436 20
8.0000	800.0	.7446-3 8	-728.1	1.7090	30.727	1.00353 -	1.0636	0.3740	1.2378	517.6	1.00000 0 1	.0000 0 1	.0000.0
8.0000	1000.0	2.9855-3 3	-663.6	1.7811	30.622	-1.00001 -	1.0002	0.3099	1.2650	586.0	0.0000.0	0 00000.0	0.0000.0
8.0000	1200.0	2.4869-3	-599.8	1.8392	30.609	-1.00007 -	1.0017	0.3184	1.2570	640.1	0.0000.0	0.0000.0	0.0000.0
8.0000	1400.0	2.1266-3	-531.8	1.8915	30.538	-1.00213	1.0397	0.3707	1.2309	635.0	0.0000.0	0.0000.0	0.0000.0
8.0000	1600.0	1.8482-3	-454.9	1.9430	30.331	-1.00024	1.0036	0.3293	1.2502	740.5	0.0000.0	0.0000.0	0.0000.0
8.0000	1800.0	1.6424-3	-338.9	1.9318	30.323	-1.00020	1.0018	0.3316	1.2472	784.6	0.0000.0	0.0000.0	1.857 18
8.0000	2000.0	1.4778-3	-321.9	2.0171	30.315	-1.00029	1.0034	0.3393	1.2410	825.1	9.8553-2	7.0164-2	8.816 18
8.0000	2200.0	1.3426-3	-252.1	2.0503	30.295	-1.00065	1.0135	0.3635	1.2265	860.5	3.8965-1	7.7845-2	3.128 19
		÷				-1.00239					÷	œ.	œ.
8.0000	2600.0	1.1234-3	-65.8	2.1276	29.929	-1.00613	1.1499	0.6010	1.1625	915.9	3.1884 0	9.2925-2	2.145 20
8.0000	2800.0	1.0276-3	70.3	2.1779	29.513	-1.01119	1.2565	0.7585	1.1480	951.6	7.3816 0	1.0022-1	4.603 20
8.0000 8.0000 8.0000	3000.0	9.3850-4	236.8	2.2353	28.88	-1.01716	1.3660	0.9037	1.1427	993.3	1.5052 1	1.0719-1	8.777 20
P, ATM													

THERMAL PROPERTIES

000	4-4	4.9	025	414	222	795	955	452	2.1	ф 0 ф	4-1	ଯ
6.000	7.681	ø	2.5	23	-1.01	1.2	0.7	1.1	95	9.039	1.335	4.254
6.0000	7.0050-4	259.4	2.2627	28.740	-1.01843	1.3912	0.9410	1.1413	995.2	1.8582 1	1.4232-1	8.161 20
7.0000	3.2750-3	-727.6	1.7181	30.713	-1.00331	1.0596	0.3698	1.2394	518.1	0.000.0	0.0000.0	0.0000.0
7.0000	2.6123-3	-653.6	1.7858	30.622	-1.00001	1.0002	0.3099	1.2650	586.1	0.0000.0	0.0000.0	0.0000.0
7.0000	2.1760-5	-599.8	1.8479	30.609	-1.00008	1.0019	0.3188	1.2558	640.0	0.0000.0	0.0000.0	0.0000.0
7.0000	1.8503-3	-531.3	1.9006	30.529	-1.00235	1.0439	0.3759	1.2290	634.5	0.000.0	0.0000.0	0.0000.0
7.0000	1.6171-3	-454.8	1.9517	30.330	-1.00023	1.0034	0.3290	1.2504	750.6	0.0000.0	0.0000	0.0000.0
7.0000	1.4371-3	-388.9	1.5906	30.322	-1.00021	1.0019	0.3316	1.2471	784.6	0.0000.0	0.00000	1.789 18
7.0000	1.2930-3	-321.8	2.0259	30.314	-1.00031	1.0035	0.3397	1.2408	825.0	1.0903-1	8.0232-2	8.498 18
7.0000					Υ.						٠.	×
7.0000	1.0735-3	-170.6	2.0945	30.202	~1.00258	1.0650	0.4619	1.1890	886.3	1.3308 0	9.7649-2	8.518 19
7.0000	9.8214-4	-62.2	2.1378	29.933	-1.00645	1.1579	0.5146	1.1606	915.5	3.5220 0	1.0624-1	2.072 20
7.0000 7.0000 7.0000 2800.0 2600.0 2400.0	8.9782-4	76.9	2.1893	29.468	-1.01166	1.2670	0.7754	1,1467	951.8	8.1362 0	1.1451-1	4.441 20
P, ATM T, DEG K	RHO, KG/CU M	H, CAL/G	S, CAL/(G)(K)	M, MOL WI	(DLV/DLP)T	(DLV/DLT)P	C, CAL/(6)(K)	GAMMA (S)	SON VEL M/SEC	SIGMA MHO/M	BETA	ELEC NUM DEN

THERMODYNAMIC EQUILIBRIUM PROPERTIES AT ASSIGNED TEMPERATURE AND PRESSURE

P, ATM	6.0000 6.0000 6.0000	6.0000	6.0000		6.0000	6.0000	6.0000	6.0000	6.0000	6.0000	5.0000	5.0000	5.0000
T, DEG K	2600.0	2400.0	2200.0	2000.0	1800.0	1600.0	1400.0	1200.0	1000.0	800.0	3000.0	2800.0	2600.0
RHO, KG/CU M	8.4097-4	9.1978-4	1.0067-3	÷	1.2317-3	1.3361-3	1.5939-3	1.8651-3	2.2391-3	2.8053-3	5.8174-4	6.3856-4	9-0666.9
H, CAL/G	-57.9	-159.0	-251.5		-333.8	-454.8	-530.6	-599.7	-653.6	-727.2	274.5	94.8	-52.4
S, CAL/(6)(K)	2.1497	2.1053	2.0595		2.0007	1.9619	1.9111	1.8579	1.7998	1.7287	2.2803	2.2184	2.1639
M, MOL HT	29.903	30.139	30.239		30.321	30.329	30.517	30.609	30.622	30.698	28.641	29.347	29.864
(DLV/DLP)T	-1.0068E	-1.00232	-1.00077	ì	-1.00022	-1.00023	-1.00265	-1.00009	-1.00001	-1.00302	-1.01934	-1.01290	-1.00733
(DLY/DLT)P	1.1674	1.0711	1.0162		1.0020	1.0032	1.0455	1.0022	1.0001	1.0543	1.4097	1.2947	1.1791
C, CAL/(6)(K)	0.6307	0.4730	0.3687		0.3317	0.3236	0.3829	0.3193	0.3099	0.3640	0.9693	0.8201	0.6505
GAMMA (S)	1.1584	1.1850	1.2239		1.2471	1,2595	1.2265	1.2565	1.2650	1.2417	1.1401	1.1435	1, 1559
SON YEL MYSEC	915.1	885.4	859.7		734.6	740.6	654.0	640.0	585.1	518.7	956.4	952.4	914.7
SIGMA MHO/M	3.9464 0	1.4896 0	4.7999-1	÷	0.0000.0	0.0000.0	0.0000.0	0.0000.0	0.0000.0	0.0000	2,1379 1	1.0339 1	4.5079 0
BETA	1.2400-1	1.1399-1	1.0392-1	o,	0.0000.0	0.0000.0	0.0000.0	0.0000	0.0000.0	0.0000.0	1.7003-1	1.6010-1	1.4835-1
ELEC NUM DEN	1.989 20	8.168 19	2.887 19	∞	1.714 18	0.0000.0	0.0000.0	0.0000.0	0.0000.0	0.0000.0	7.859 20	4.036 20	1.893 20

ROPERTIES	
THERMAL	

4.0000 2400.0 .1240-4 .1240-4 2.1342 30.151 1.00553 1.0895 0.5053 1.1779 882.9 882.9 .9951 281.19
4,0000 2600.0 5.587-4 6 2.1815 29.813 -1.00794 - 1.1940 1.1529 914.3 5.2922 1
4.0000 2800.0 2800.0 107.7 2.2331 29.260 1.3142 0.8516 1.1415 953.0 1.2062 1.9978-1
4.0000 3000.0 4.6333-4 28.314-2.3024 -1.02049 1.4332 1.0055 1.1333 998.1 2.5214 1
5.0000 800.0 2.3370-3 1.726.7 30.682 1.06252 1.0472 0.356.1 1.2472 0.0000 0.0000
5.0000 1.8656-3 -653-3 1.813:5 1.813:5 -1.00100 1.0126 0.3200 1.2510 585.2 0.0000 0.0000
5.0000 1.5542-3 1.5542-3 1.5542-3 30.603 1.0027 1.2551 639.9 0.0000 0
5.0000 1.3276-3 1.3276-3 1.523-7 1.0335 1.0578 1.0578 1.2231 6.83.2 0.0000 0.0000
5.0000 1600.3 1.1550-3 1.9739 30.328 -1.00023 1.0030 0.3283 1.2599 740.7 0.0000 0.0000
5.0000 1.0264-3 1.0264-3 2.0127 30.320 1.0023 1.0021 1.2471 784-6 0.0000 0.0000 1.628 18
5.0000 2000.0 2000.0 -2243.4 -321 2.043.3 1.0063 0.3607 1.2602 824.9 1.397-1 1.1269-1 7.732 18
5.0000 2200.0 8.3830-6 -251.1 2.0315 50.284 -1.06685 1.0181 0.3726 1.2221 852.1 857.1 1.2480-1
5.0000 2400.0 2400.0 2.6607-4 -166.8 -251.1 2.182 2.0315 30.173 1.0730 1.0730 1.0730 1.1824 1.2221 884.3 884.3 884.3 1.2591 1.2591 1.2591 1.2591 1.2591 1.2589-1
P, ATH T, DEG K RHO, KG/CU M H, CAL/G S, CAL/(G)(K) M, MOL WT (DLV/DLP)T (DLV/DLT)P C, CAL/(G)(K) GANMA (S) SON VEL M/SEC SIGMA MHO/M BETA ELEC NUM DEN

THERMODYNAMIC EQUILIBRIUM PROPERTIES AT ASSIGNED TEMPERATURE AND PRESSURE

0 3.0000 73.0000 0 2600.0 2400.0	4 4.1822-4 4.5830-4 5.	4.44[- [.47- 4	5 -25.1 -159.4 1 2.2047 2.1550 9 29 742 30 118	5	5 - 55.1 - 159.4 1 2.2047 2.1550 29.742 30.418 3 -1.0037 -1.00411 -1	5 - 55.1 - 159.4 1 2.2047 2.1550 29.742 30.118 3 -1.00377 -1.00411 -1 6 0.7104 0.5538	55.1 -159.4 1 2.2047 2.1550 29.742 30.118 3 -1.00377 -1.00411 -1 1.2142 1.1048 6 0.7104 0.5538 0 1.1495 1.1720	2.2047 2.1550 2.2047 2.1550 29.742 30.118 3 -1.00577 -1.00411 -1 1.2142 1.1048 6 0.7104 0.5538 0 1.1493 1.1720 9 914.0 851.2	125.6 -55.1 -159.4 -249.5 2.2641 2.2047 2.1550 2.1159 29.739 29.742 30.118 36.269 -1.01493 -1.00377 -1.00411 -1.00112 1.3406 1.2142 1.1048 1.0251 0.8946 0.7104 0.5338 0.3863 1.1350 1.1493 1.1720 1.2158 953.9 914.0 851.2 857.1 1.4628 1 6.4804 0 2.4477 0 7.8591-1
_	3 3.4538-4 2 321.2	101.4	;	_	^^	_	٠.	M	0
	-3 1.8586-							0	c
	-3 1.4922-3 .7 -663.3	~ ~	1		_	. +-	_	0	0
	-3 1.2434-34-34-34-34-34-34-34-34-34-34-34-34-	· ~			٠,	۸,		0	c
	-4 1.0612-3 6 -523.4	10.4		m	Δ.	,	m	0	
	4 9.2395-4		- 1					ö	c
4.0000 4.0000 4.0000 2200.0 2000.0 1800.0	4 8.2107 5 -333	2.027	8 -1.003 -1.003	1.000	16 0.33	78 1.24	7 784	-1 0.0000	0000
30 4.00(.0 2000	-4 7.3873 -4 -321	2.06	36 -1.000	1.00	30 0.34	35 1.239	.3 824	-1 1.6333	4 4 6077
4.00(6.7091- -250	25.09	-1.000	1.02	0.378	1.219	858	6.4152	1 EC 1C
P, ATM T, DEG K	RHO, KG/CU H H, CAL/G	S, CAL/(G)(K)	(DLV/DLP)T	(DLV/DLT)P	C, CAL/(G)(K)	GAMMA (S)	SON VEL M/SEC	SIGNA MHO/M	BCTA

THERMAL PROPERTIES

0000 1010 1010 1010 1010 1010 1010 101
2.0000 2.5000.0 2.1084 2.1084 1.0064 1.0064 1.2345 2.6554-1 2.8258-1
2.0000 2.200.0 3.5518-4 -247.8 2.1453 30.254 -1.00139 1.0326 1.2096 855.2 1.0420 1.0420 3.1331-1
2.0000 2400.0 3.0550-4 -151.9 2.1849 30.062 -1.00535 1.1297 1.1537 878.9 3.2430 3.4521-1 5.915 19
2.0000 2.775-4 2.735-6 -18.9 2.2380 29.629 -1.01001 1.2447 1.7444 913.8 8.5464 0 3.7144-1
2.5000 2.5203-6 153.6 2.5018 2.5018 28.953 -1.01671 1.3505 0.1358 955.6 3.9535-1
2.0000 3000.0 3000.0 2.2809-4 2.3741 28.074 -1.62451 1.5141 1.1310 1.1350 4.0172 6.130 20
3.0000 800.0 1.4007-3 -725.7 1.7754 30.650 -1.00149 1.2553 1.2553 522.1 0.0000 0
3.0000 1.1192-3 -663.3 1.8450 -1.00000 0.0000 0.0000 0.0000 0.0000 0.0000 0.0000 0.0000 0.0000 0.0000 0.0000
3.0000 1200.0 9.3247-4 -559.6 1.9031 30.606 0.3227 1.2544 639.5 0.0000 0
3.0000 1400.0 7.9494-4 -525.3 1.9594 30.440 -1.00492 0.4925 0.4925 1.2107 680.4 0.0000 0
3.0000 1600.0 4.6.9292-4 7.455.6 5.0074 5.1.00024 4.1.0027 4.1.0027 6.740.9 0.0000 0
3.0000 1800.0 6.1576-4 -358.7 2.0462 30.316 -1.00027 0.3024 0.3024 0.3024 0.3024 1.2470 784.6 0.0000 0.0000 0.0000
3.0000 3.0000 5.5398-4 6.1576-4 -321.3 -338.7 2.0817 2.0462 30.305 30.316 -1.00042 -1.00027 1.0054 0.3321 1.2390 1.2470 824.5 784.6 2.0043-1 0.0000 0 1.8797-1 0.0000 0 6.665 18 1.405 18
P, ATM T, DEG K RHO, KG/CU H H, CAL/6 S, CAL/(6)(K) H, MOL HT (DLV/DLP)T (DLV/DLP)T (DLV/DLT)P C, CAL/(6)(K) GAMNA (S) SON VEL M/SEC SIGMA MHO/M BETA ELEC NUM DEN

THERMODYNAMIC EQUILIBRIUM PROPERTIES AT ASSIGNED TEMPERATURE AND PRESSURE

1.0000	2.0518-4	2, 1185	30,305	-1.00033	1.0034	0.3332	1.2455	784.6	0.0000 0	0.0000.0	1.007 18
1.0000	1.8456-4	2, 1543	30.288	-1.00061	1.0094	0.3506	1.2347	823.3	4.2950-1	5.6750-1	4.730 18
1.0000	1.6740-4	2.1907	30.219	-1.00203	1.0507	0.4368	1,1964	851.0	1.6677 0	6.2841-1	1.659 19
1.0000	1.5207-4	2.2374	29.947	-1.00621	1.1613	0.6382	1,1547	877.2	5.1844 0	6.8647-1	4.720 19
1.0000	1.3781-4	2.2974	29.400	-1.01238	1.3032	0.8643	1,1369	914.3	1.3365 1	7.3599-1	1, 133, 20
1.0000	1.2442-4	2.3697	28.536	-1.02018	1.4579	1.0879	1.1307	9265	2.8689 1	7.7423-1	2.316 20
1.0000	1.1154-4	2.4513	27.557	-1.02931	1.6085	1.2794	1.1319	1012.2	5.9911 1	7.8206-1	4.788 20
2.0000	9.3338-4	1.8022	30.635	-1.00031	1.0145	0.3191	1.2633	523.7	0.0000.0	0.0000.0	0.0000.0
2.0000	7.4612-4	1.8714	30.611	-1.00000	1.0001	0.3638	1.2651	536.2	0.0000.0	0.0000.0	0.0000.0
2.0000	•			~						•	
2.0000	5.2898-4 -523.1	1.9882	30.384	-1.00115	1.0250	0.3563	1.2372	688.4	0.0000.0	0.0000.0	0.0000.0
2.0000 1600.0	4.6190-4 -454.4	2.0341	30.321	-1.00025	1.0026	0.3272	1.2517	741.0	0.0000.0	0.0000.0	1.729 17
2.0000 2.0000 2.0000 1800.0 1600.0 1400.0	4.1046-4 -388.5	2.0729	30.312	-1.00031	1.0028	0.3324	1.2469	784.6	0.0000.0	0.0000.0	1.246 18
P, ATM T, DEG K											

THERMAL PROPERTIES

0.5000 1600.0 1543-4 -454.0 2.1552 30.309 1.00035 1.2521 741.3 168, 17)
0.5000 180.0 1.0256-4 -3.88.0 2.58.0 30.296 1.0042 0.3342 1.2461 784.6 0.0000 0 0	
0.5000 2000.0 2.234-5 -319.3 2.273 30.273 30.273 1.0137 0.3595 1.2297 821.9 821.9 821.9	
0.5000 2200.0 8.3555-5 -237.3 2.237.3 30.157 1.0725 0.4838 1.1509 846.2 846.2 2.6278 0.1.25910	
0.5000 2400.0 7.5661-5 2.2520 29.800 -1.00797 1.2092 0.7270 1.1444 8.0745 0.7270 1.3575 1.3575 1.3575	\- \- \- \- \- \- \- \- \- \- \- \- \- \
0.5000 2600.0 6.8251-5 2.350.4 29.122 29.122 1.3712 0.9835 1.1303 916.0 1.4493	:
0.5000 2800.0 6.1263-5 2.4426 28.1516 28.1516 1.5451 1.2308 1.1270 965.4 4.2799 1.4903	2
0.5000 300.00 300.00 5.4740-5 2.5353 26.9511 -1.03501 1.4540 1.1295 1.1295 1.1295 1.1295 1.4760 1.4760 1.4760 1.4760 1.4760 1.4760	}
1.0000 800.0 800.0 1.8476 1.8476 30.626 1.00024 1.2701 525.2 0.0000 0.0000	
1.0000 3.7306-4 -663.3 1.916-4 50.611 1.0001 1.0001 1.2651 586.2 0.0000 0.0000	
1.0000 1200.0 1.003-4 1.9753 30.585 -1.00101 1.0197 0.3420 1.2445 637.2 0.0000 0.0000	
1.0000 1400.0 2.6416-4 2.0353 30.347 -1.00254 1.0558 0.4017 1.2183 683.6 0.0000 0	
1.0000 1.0000 1.0000 1.0000 1.0000 1.0000 1.0000 1.2500 1.454.2 -520.6 -598.6 2.0353 1.9753 30.316 30.347 30.585 -1.00029 -1.0028 1.0558 1.0197 0.3270 0.4017 0.3420 1.2520 1.2183 1.2445 7.0000 0.000	
P, ATM T, DEG K RHO, KG/CU M H, CAL/G S, CAL/(G)(K) M, HOL MT (DLV/DLP)T (DLV/DLP)T C, CAL/(S)(K) GANNA (S) SON VEL M/SEC SIGMA MHO/M BETA	

THERMODYNAMIC EQUILIBRIUM PROPERTIES AT ASSIGNED TEMPERATURE AND PRESSURE

Ö	۵	2.3325-4	7	Ψ.	33	-1.0	÷.	ö	÷	щ	0.00	0.0	0.00
		1.8653-4				7					ö	ö	Ö
0.5000	1200.0	1.5519-4	-597.3	2.0214	30.552	-1.00139	1.0296	0.3580	1.2363	635.3	٦,	0.0000.0	٦,
0.5000	1400.0	1.3199-4	-519	2.08	30.3	-1.000	1.00	0.32	1.25	693	0.000	3.000	0.000
P, ATM	T, DEG K	RHO, KG/CU M	H, CAL/G	S, CAL/(G)(K)	H, HOL HT	(DLV/DLP)T	(DLV/DLT)P	C, CAL/(6)(K)	GANNA (S)	SON VEL M/SEC	SIGMA MHO/M	BETA	ELEC NUM DEN

发	8.772E+00 4.6034E+00 8.8237E-01 3.1235E-01 8.8128E-01 9.0000E+00 0.0000E+00	.0000E+0
SIG	7.3816E+00 3.884E+00 3.884E+00 3.885E+00 0.0000E+00 0.000E+00 0.000E+00 0.000E+00 0.0000E+00 0.0000E+00 0.0000E+00	.0000E
NO TOT	1.6431E+12 1.8902E+12 2.0596E+12 2.0596E+12 2.2653E+12 6.4010E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5346E+12 1.5369E+12 1.5369E+12 1.5369E+12 1.5369E+12 1.5369E+12 1.5369E+12 1.5429E+12 1.5429E+12 1.5465	. 85 19E + 0 . 1953E + 0
GAM	1.1427E+00 1.1480E+00 1.255E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.2572E+00 1.255E+00 1.255E+00 1.255E+00 1.255E+00 1.256E+00 1.255E+00 1.255E+00 1.256E+00 1.2571E+00 1.250E+00 1.2	+ +
G G	3.7813E+03 2.5146E+03 1.521EE+03 1.521EE+03 1.527EE+03 1.527EE+03 1.527EE+03 1.527EE+03 1.527EE+03 1.526E+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03 1.538SE+03	.5555E+0
RHO	9.3830E-01 1.0276E+00 1.0276E+00 1.572E+00 1.5426E+00 2.455E+00 2.455E+00 2.455E+00 3.7446E+00 3.7446E+00 3.7446E+00 3.7446E+00 1.555E+00 1.2930E+00 1.2930E+00 1.4371E+00 1.4371E+00 1.4371E+00 1.4371E+00 1.4371E+00 1.4371E+00 1.617	.3370E+0
ဟ	9.3525E+03 8.782E+03 8.782E+03 8.782E+03 8.783E+03 8.2919E+03 8.2919E+03 7.4521E+03 7.4521E+03 7.4521E+03 7.4521E+03 7.4521E+03 7.4521E+03 7.4521E+03 8.445E+03 8.4466	.2647E+0
æ	2.9982E+05 1.1.3458E+05 1.34	Z.//5%E+ 3.0405E+
至	2.95138 # 01 2.95138 # 01	.0632E+01
-	22.22.6.0000000000000000000000000000000	.0000E+
ο.	8.00000E+000	.0000E+0

7.77.74.00 7.77.75.00 7.77.7 2.5248E+01 1.2062EE+01 1.5252EE+01 1.5252EE+01 1.65252EE+01 1.65252 8.3292E+11 9.4537E+11 1.0270E+12 1.2536E+12 1.2536E+12 1.2536E+12 5.0203E+02 6.3104E+12 6.3104E+12 6.3104E+12 6.3104E+12 6.3104E+12 6.3104E+12 6.3104E+12 7.6536E+02 6.3104E+11 7.6536E+02 6.3266E+02 6.3366E+02 7.6450E+11 7.6536E+02 6.3366E+02 6.336E+02 6.3366E+02 6.3366E 1.1509 1.1519 1.21959 2.55235E+93 2.55235E+93 2.55235E+93 2.55235E+93 2.55235E+93 2.5525E+93 2.5535E+93 2.5535 5.6533 25.52E+03. 25.52E+03. 25.52E+03. 25.52E+03. 25.52E+03. 25.52E+03. 25.52E+03. 25.53E+03. 25.53E 1.2307F+05 4.5055F+05 1.0478F+05 1.0478F+06 1.0478 8514E+0101253E+01101253E+0 ----- 8 W O O O O O O O

4.7831E+00 1.1334E+00 4.7203E+00 1.6557E+01 1.6557E+01 0.0000E+00
5.9911E+01 1.3355E+01 1.6577E+00 1.6577E+00 0.0000E+00
2.2691E+11 2.2691E+11 2.3645E+11 2.5632E+11 3.6460E+12 7.6460E+02 6.355E+02 6.355E+02 6.355E+02 6.355E+02 1.2675E+11 1.26
1.1319 1.13529 1.13529 1.13529 1.13529 1.13529 1.13529 1.13539
5.3530E+03 3.6162E+03 2.6764E+03 1.65277E+03 1.65277E+03 1.6527E+03 1.6528E+03 1.6538E+0
1.17946 1.27456 1.5207
1.0256#+05 9.5125#+03 9.5125#+03 9.557#+03 8.553#+03 8.553#+03 8.555#+03
1.8576 # 105
2.7557E+01 2.8528E+01 2.9400E+01 3.0219E+01 3.0219E+01 3.0316E+01 3.0316E+01 3.0316E+01 3.0316E+01 3.0316E+01 3.0328E+01 2.6951E+01 2.6951E+01 2.6951E+01 2.6951E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01 3.0329E+01
3.0000E+03. 2.5000E+03. 2.5000E+03. 3.0000
1.00000E 1.0000E 1